Stability Indicating RP-HPLC PDA Analytical Method Development and Validation for the estimation of Besifloxacin Hydrochloride in Bulk and Formulation

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ABSTRACT

The goal of the study was to provide an overview of the technique development and validation of a stability-indicating HPLC approach for routine analysis of besifloxacin hydrochloride in pharmaceutical product (ophthalmic suspension). HPLC method was development on octadecasilyl silica (C_{18} , 250 mm x 4.6 mm x 5u) column at 37 °C and isocratic mode with a running solvent (phosphate buffer: methyl alcohol 40:60 % V/V) using flow rate (1.0 mL minute-1) and UV wavelength 292 nm. Proposed method was validated for specificity, linearity, accuracy, precision, range and robustness according to ICH Q2 (R1) standards. The collected results attest to the validated method's compliance with the set acceptance standards. Besifloxacin hydrochloride was subjected to hydrolytic, oxidative, thermal and photolytic stress conditions. These samples were then examined using our suggested approach. Hence, this method can be used for routine use for determination assay of besifloxacin hydrochloride drug substance (API) and drug product (ophthalmic suspension).

Keywords: Besifloxacin hydrochloride (BFHC), Stress testing, Validation, Photo-diode Array (PDA) and drug product (Ophthalmic suspension)

INTRODUCTION

Besifloxacin hydrochloride (BFHC) is a 4th generation fluoroquinolone antibiotic used as an ophthalmic formulation to prevent infectious complications in patients receiving laser treatment for cataracts, bacterial conjunctivitis brought on by aerobic, facultative grampositive microorganisms, and aerobic and facultative gram-negative microorganisms. Staphylococcus aureus, Staphylococcus epidermidis, Staphylococcus hominis, Streptococcus mitis, Streptococcus oralis, Streptococcus pneumoniae, and Streptococcus salivarius are among the bacteria that can cause skin infections, etc $1-2$. It is a disease that is extremely contagious and has the potential to spread quickly. The antibiotics used to treat this pathology are crucial because they reduce the spread of the infection, time missed from work or school, infection recurrence, and the danger of further, perhaps irreversible ocular damage²⁻⁴.

Besifloxacin functions by adhering to the bacterial enzymes DNA gyrase and topoisomerase IV, which are necessary for bacterial DNA replication⁵. When these enzymes are inhibited, bacterial chromosomal DNA cannot be transcriptionally transcribed, replicated, or divided during cell division (Central Dogma)6-8. Bactericidal activity is a property of besifloxacin. The most frequent mechanisms of besifloxacin resistance include changes to the bacterial membrane outflow pumps and point mutations in the genes that code for DNA gyrase and topoisomerase IV9 . Besifloxacin, however, has certain hypothetical advantages over other fluoroquinolones in the case of resistance6-8. High affinity for DNA gyrase and topoisomerase IV, in contrast to nearly all fluoroquinolones, which only have sympathy for one of them and correlate with the occurrence of resistance¹⁰. Therefore, for a bacteria to acquire besifloxacin resistance, besifloxacin-encoding genes must undergo the evolution of voluntary mutations, which is a less frequent occurrence^{9,11}. Additionally, only topical ophthalmic usage of besifloxacin is available, which restricts systemic exposure and, as a result, slows the development of resistance¹²⁻¹³. In 2009, Bausch & Lomb

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received FDA approval for BEX. HCl, a medication used to treat non viral bacterial inflammation of the conjunctiva of the eye. BFHC is 7-[(3R)-3-aminoazepan-1-yl]-8 chloro-1-cyclopropyl-6-fluoro-4-oxo-1,4-hydroquinoline-3-carboxylic acid hydrochloride (IUPAC Name)^{1,11,14}. Structure of besifloxacin HCl, containing basic a basic ring and side side structural components as shown in Fig. 1.

Fig. 1: Structure of besifloxacin HCl

Method development was performed using different column chemistry like C_8 and C_{18} , but C_{18} column shows good resolution and peak compared to $\mathsf{C}_{_{\bf 8}}$ column $^{_{15}}$. Hence C_{18} column of two different brands and particle size, Agilent Eclipse XDB-C₁₈ column (150x4.6mm i.d., 3.5 μ m particle size) and Kromasil C₁₈ (250mm×4.6mm×5 μ) were used. The best option for producing a symmetrical peak with good resolution was a Kromasil manufacture column. To optimize chromatographic conditions, different mobile phase ratios with organic solvent trial were taken and final ratios of running phase (buffer phosphate pH 3.0: methyl alcohol 40:60 V/V) using flow rate (1.0 mL minute-1) and UV lambda max 292 nm showing highest sensitivity¹⁶. The above method validated according to International Conference on Harmonization standards and proved to be specific, rapid, accurate, precise, reproducible; robust method for assay of besifloxacin hydrochloride drug substance (API) and drug product (ophthalmic suspension)15-17.

MATERIALS AND METHODS

Details of chemicals/ reagents used for research study with their make and grade mentioned as in the Table I.

Table I: Chemicals/ reagents details

Sr. No	Chemicals/ reagents	Make	Grade
01.	Phosphate buffer	Merck	HPLC and equivalent
02.	acetonitrile	Merck	HPLC and equivalent
03.	Methanol	Merck	HPLC and equivalent
04.	Water		Milli Q

Details of product used, Batch No and making, for research study as mentioned in the Table II.

Table II: Sample details

Sr. No	Product name	Batch No.	Make
01.	Besifloxacin ophthalmic suspension 30 mg mL-1	01A110 and Ajanta 1BA821	Pharma

Details of instrument on which research carried out with their making as mentioned in the Table III.

Table III: Instrument details

Sr. No.	Instrument name	Mode	Make
01.	HPLC Shimadzu manual sampler with PDA detector	HPI C (LC20AD) and detector (LC20AD)	Shimadzu Instruments

RESULTs AND DISCUSSION

Method development and chromatographic condition optimization

Initial method development was performed on the $C_{\rm g}$ column at 0.8 mL minute⁻¹ flow using different ratios of buffer and acetonitrile, no peak elution was observed. Numerous modifications were made to the mobile phase's composition and chromatographic settings to increase the possibility that chromatographic systems' selectivity would alter¹⁵⁻¹⁷. This modification included the change like column chemistry ${\mathsf C}_{_{\mathsf B}}$ replace with ${\mathsf C}_{_{18}},$ organic modifier acetonitrile replaced by methanol, flow change to 1.0 mL minute-1, running phase ratio (buffer phosphate pH 3.0: methyl alcohol 40:60 V/V) and column oven temperature to 37 °C. Wavelength and injection volume finalize based on peak response and sensitivities at 292 nm of besifloxacin hydrochloride.

Besifloxacin HCl is slightly soluble in methanol and it is sparingly soluble in aqueous solutions. We diluted the organic solvent solution with aqueous buffers to increase aqueous solubility; as a result, the diluent was chosen as the mobile phase¹⁸.

UV spectrum of besifloxacin HCl indicating lambda max is as shown in Fig. 2.

Fig. 2: UV spectrum of besifloxacin hydrochloride

Spectrum of besifloxacin HCl indicating peak purity is shown in Fig. 3.

Fig. 3: Peak purity

Details of impurity, peak purity index, single point threshold and minimum peak purity index of besifloxacin hydrochloride used for research study are mentioned in the Table IV.

Table IV: Peak purity

Spectra of besifloxacin HCl indicating peak profile is shown in Fig. 4.

Fig. 4: Peak profile of besifloxacin hydrochloride

Final chromatographic condition follows

Preparation of buffer for mobile phase

3.4 g of potassium dihydrogen phosphate was weighed and dissolved in 1000 mL of water. Ortho phosphoric acid solution was used to adjust the pH between 3.0 \pm 0.05 and was sonicated to dissolve the gas.

Preparation of mobile phase and diluent

Mobile phase buffer and methanol was prepared in the ratio of 40:60 V/V.

Details of optimized conditions of developed method of besifloxacin hydrochloride used for research study are mentioned in the Table V.

Table V: Conditions for method development

Preparation of standard solutions

3 mg of BFHC standard was weighed and dissolved in 50 mL of mobile phase. Mobile phase mixed into it, to make up the volume 100 mL (Concentration of besifloxacin hydrochloride 30 µg mL⁻¹) and was sonicated to dissolve the gas.

Preparation of sample solutions

For drug substance (API)

3 mg of besifloxacin hydrochloride API was weighed and dissolved in 50 mL of mobile phase. Mobile phase mixed into it, to make up the volume 100 mL (Concentration of besifloxacin hydrochloride 30 µg mL-1).

For drug formulation (ophthalmic suspension)

Transferred 0.5 mL of ophthalmic suspension (Equivalent 3 mg mL-1 besifloxacin hydrochloride) in 100 mL flask, mobile phase was added into it, to make up the volume 100 mL and mixed well (Concentration of besifloxacin hydrochloride 30 µg mL-1).

Method validation

The experiments were carried out to verify the method's validity, and based on the results, the method's suitability for its intended use¹⁹⁻²⁰.

The following parameters were validated by the aforementioned analytical assay method;

- 1. System suitability / system precision
- 2. Specificity (diluent interference and forced degradation study)
- 3. Method precision
- 4. Accuracy
- 5. Detection Limit
- 6. Linearity
- 7. Range
- 8. Robustness

System suitability and system precision

System suitability testing is a vital component of the analytical method that entails injecting five replicated injections of standard solutions into the system, % RSD-

Fig. 5: Chromatogram for working standard of besifloxacin hydrochloride Fig. 6: Blank chromatogram

1.5 (should be less than 2.0), tailing factor-1.548 (should be less than 2.0) and plate count-6002.088 (should be more than 2000) of main peak. Chromatogram of besifloxacin HCl indicating, retention time for working standard is shown in Fig. 5.

Details of standard peak (retention time, peak area, theoretical plate and tailing factor) of besifloxacin hydrochloride used for research study are mentioned in Table VI.

Details of system suitability parameters (ret. time, USP resolution (Rs), theoretical plates, no. of theoretical plates (N), capacity factor (k' prime) and tailing factor (T)) of besifloxacin hydrochloride meeting criteria are mentioned in Table VII.

Table VII: System suitability parameters

Sr.	Parameter	Besifloxacin
No.		peak
01.	Retention time (R.)	6.917 minutes
02.	USP resolution (Rs)	1.69
03.	Tailing factor (T)	1.548
04.	No. of theoretical plate (N)	6002.088
05.	Capacity factor (k' prime)	1.46

Specificity

At the retention time of the main peak from the diluent, no peaks were seen and all disturbances and peak purity cleared. Blankchromatogram indicating, zero interference as shown in Fig. 6.

Details of forced degradation study (acid stress, base stress, oxidation stress, thermal stress and U V stability) are mentioned in the Table VIII.

Table VIII: Details of forced degradation study

Chromatogram indicating stability of besifloxacin hydrochloride for acid stress; 1 N HCl at 40 °C heat, for 30 minutes as shown in Fig. 7.

Fig. 7: Chromatogram for 1 N HCl at 40 ºC, for 30 minutes

Details of acid stress study of besifloxacin hydrochloride; 1 N HCl at 40 °C, for 30 minutes are mentioned in Table IX.

Chromatogram indicating stability of besifloxacin hydrochloride for 1 N HCl at 40 °C heat, for 60 minutes is shown in Fig. 8.

Fig. 8: Chromatogram for 1 N HCl at 40 ºC, for 60 minutes

Details of acid stress study of besifloxacin hydrochloride; 1 N HCl at 40 °C, for 60 minutes are mentioned in Table X.

Chromatogram indicating degradation (base stress) study, of besifloxacin hydrochloride; 30 minutes with 0.1 M NaOH at 40 $\,^{\circ}$ C is shown in Fig. 9.

Details of degradation (base stress) study of besifloxacin hydrochloride; 0.1 M NaOH at 40 \degree C, for 30 minutes are mentioned in Table XI.

Fig. 9: Chromatogram for 30 minutes with 0.1 M NaOH at 40 oC

Chromatogram indicating stability (base stress) of besifloxacin hydrochloride for 45 minutes with 0.1 M NaOH at 40 $\,^{\circ}$ C heat is shown in Fig. 10.

Fig. 10: Chromatogram for 45 minutes with 0.1 M NaOH at 40 oC heat

Details of degradation study (base stress) of besifloxacin hydrochloride; 0.1 M NaOH at 40 °C, for 45 minutes as mentioned in the Table XII.

Chromatogram indicating stability (base stress) of besifloxacin hydrochloride for 45 minutes with 0.1 M NaOH at 40 \degree C heat is shown in Fig. 11.

Fig. 11: Chromatogram for 60 minutes with 0.1 M NaOH at 40 oC heat

Details of degradation study (base stress) of besifloxacin hydrochloride; 0.1 M NaOH at 40 °C, for 60 minutes are mentioned in Table XIII.

Table XIII: Peak retention times and peak areas after 60 minutes of heating with 0.1 M NaOH at 40 oC

Chromatogram indicating stability (base stress) of besifloxacin hydrochloride for 1 h 30 minutes with 0.1 M NaOH at 40 \circ C is shown in Fig. 12.

Fig. 12: Chromatogram for 1 h 30 minutes with 0.1 M NaOH at 40 oC heat

Details of degradation study (base stress) of besifloxacin hydrochloride; 0.1 M NaOH at 40 °C, for 1 h 30 minutes is mentioned in Table XIV.

Table XIV: Peak retention times and peak areas after 1 h 30 minutes of heating with 0.1 M NaOH at 40 oC

Sr. No.	Ret. time (Minutes)	Peak area
01.	3.023	10674
02.	3.280	6825
03.	4.516	74167
04.	7.274	2591
05.	7.113	601444

Chromatogram indicating stability (peroxide stress) of besifloxacin hydrochloride for 3% $H_2O_2^{}$ 40 °C, for 30 minutes is shown in Fig. 13.

Fig. 13: Chromatograms for 3% H₂O_{2,} 40 °C, for 30 minutes

Details of forced degradation study (peroxide stress) of besifloxacin hydrochloride; 3% H_2O_2 at 40 °C, for 30 minutes are mentioned in the Table XV.

Chromatogram indicating stability (peroxide stress) of besifloxacin hydrochloride for 3% H_2O_2 40 °C, for 60 minutes is shown in Fig. 14.

Fig. 14: Chromatograms for 3% H_2O_2 40 °C, for **60 minutes**

Details of forced degradation study of besifloxacin hydrochloride; 3% H₂O₂ at 40 °C, for 60 minutes are mentioned in the Table XVI.

Chromatogram indicating stability (peroxide stress) of besifloxacin hydrochloride for 3% H_2O_2 40 °C, for 1 h 30 minutes is shown in Fig.15.

Details of degradation study (peroxide stress) of besifloxacin hydrochloride; 3% H₂O₂ at 40 °C, for 1 h 30 minutes are mentioned in Table XVII.

Fig. 15: Chromatogram for 3% H₂O_{2,} 40 °C, **for 1 h 30 minutes**

Table XVII: Peak retention times & peak areas for 3% H2 O2, 40 ºC, for 1 h 30 minutes

Sr. No	Ret. Time (Minutes)	Peak area
01.	2.968	208582
02.	3.914	7624
03.	4.128	4107
04.	5.620	7034
05.	7.388	773342

Chromatogram indicating stability (thermal stress) of besifloxacin hydrochloride for 60 °C heat, for 30 minutes is shown in Fig.16.

Details of degradation study (thermal stress) of besifloxacin hydrochloride; at 60 °C heat, for 30 minutes are mentioned in the Table XVIII.

Table XVIII: Peak retention times & peak areas at 60 ºC heat, for 30 minutes

Chromatogram indicating stability (Thermal Stress) of besifloxacin hydrochloride; 60 \degree C heat & 60 minutes is shown in Fig. 17.

Fig. 17: Chromatogram for degradation at 60 ºC, for 60 minutes

Details of degradation study (thermal stress) of besifloxacin hydrochloride; 60 °C heat, for 60 minutes are mentioned in Table XIX.

Table XIX: Peak retention times & peak areas for 60 ºC, for 60 minutes

Chromatogram indicating stability (photolytic stress) of besifloxacin hydrochloride; U V chamber at 292 nm, for 24 h is shown in Fig. 18.

Fig. 18: Chromatogram for U V chamber at 292 nm, for 24 h

Details of degradation study (photolytic stress) of besifloxacin hydrochloride; UV chamber at 292 nm, for 24 h are mentioned in Table XX.

Table XX: Peak retention times & peak areas for U V chamber at 292 nm, for 24 h

Method precision

Process represents the closeness of scatter between the results. Six different sample solutions were prepared for API and ophthalmic suspension, calculated % Assay was well within acceptance criteria and % RSD was found to be less than 2.0.

Details of system suitability parameters of besifloxacin hydrochloride are mentioned in Table XXI.

Table XXI: System suitability parameters

Table XXII: Accuracy (recovery) of besifloxacin hydrochloride

Table XXIII: Linearity, range and method sensitivity

Accuracy

For getting accuracy**,** added known quantity of besifloxacin and calculated the recovery at 80%, 100% and 120% level.

Details of accuracy (recovery) of besifloxacin hydrochloride are mentioned in the Table XXII.

Linearity

Obtained test results are directly proportional to the quantity of determinands in the sample.

To check out the linearity range, serial dilutions of besifloxacin hydrochloride at 10 µg mL⁻¹, 20 µg mL⁻¹, 30 µg mL-1, 40 µg mL-1 and 50 µg mL-1 respectively were used in mobile phase and injected.

Details of linearity, for range of 10 ug mL⁻¹ to 50 ug mL-1 and method sensitivity of besifloxacin hydrochloride are mentioned in the Table XXIII.

Graph indicating linearity curve, value of R^2 and y of besifloxacin hydrochloride is shown in Fig. 19.

Fig. 19: Linearity curve

Detection limit (LOD)

During the practical lowest amount of analyte detected by, $LOD = (3.3 \times \sigma)/b$. Calculated values were confirmed by injection.

Range

A range for this method was established with adequate level of precision, accuracy, and linearity.

Robustness

Developed method is robust with respect to assess its propensity to persist unaltered by subtle but deliberate changes in the variables.

Details of robustness study of besifloxacin hydrochloride with respect to change in flow rate, Buffer pH, organic ratio, Column temperature and Wavelength on acceptance requirements of system suitability parameters (Tailing factor, Theoretical Plates, % RSD etc) as mentioned in the Table XXIV.

Table XXIV: Robustness study

CONCLUSION

Result of each individual method validation study meets its predefined acceptance criteria of ICH guidelines, which is described in respective study. It is determined that the analytical approach is precise, specific, linear, tough (rugged), and resilient (robust) based on the analytical data and findings of each investigation. As a result, the RP-HPLC method described above, which indicates stability, is appropriate for "Determination of assay of besifloxacin hydrochloride in drug substance (API) and drug product" (ophthalmic suspension). Due to the fact that all peaks are clearly separated, the procedure is very straightforward, quick, and economical, making it particularly appropriate for routine quality control analytical work.

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